



Study of Some Types of Bacteria and application of biological activity for some new Heterocyclic Compounds containing nitrogen Derivatives

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Abstract:

This broadside includes a preparation of certain bioheterocyclic multifaceted derivatives starting: 6-bromo-2-hydroxy naphthaldehyde, by the reaction between 2-amino-5-bromo pyrimidine with 6-bromo-2-hydroxy naphthaldehyde, to produce a Schiff base derivative (S). The latter compound was used as an important substance to prepare six new heterocyclic derivatives that have a pyrimidine ring. Also, newly synthesized compounds were characterized by spectral data (FTIR, ¹H NMR, ¹³C NMR), and all the prepared derivatives were studied on four types of bacteria (*Staphylococcus aureus* and *Escherichia coli*, *Enterococcus faecalis*, *Pseudomonas aeruginosa*), by paper technique disks. Biological capacity tests for these derivatives revealed favorable findings.

Keywords: Bacteria, microwave irradiation, Schiff bases, oxazepine, Imidazolidine derivatives.

Introduction:

Pyrimidine is widely found in nature as a component of nucleic acids (cytosine, thymine, and uracil) and many other natural and synthetic compounds, including pharmaceuticals, and has become known over time as a powerful pharmacophore (1). Also, Pyrimidine can be considered a cyclic amine (2). Pyrimidine is also known as 1,3-diazine. It is the fundamental substance of a large group of heterocyclic compounds and plays a vital role in many biological processes (3). Schiff bases are an important group of organic medicinal chemical compounds. They contain an azomethine group (-CH=N-) (4) and are prepared from the condensation that occurs between a Primary aromatic or aliphatic amine with an active carbonyl group, either aldehydes or ketones (aliphatic or aromatic) (5). They have been used to treat cancers, including skin cancer and brain cancer (6,7). They are also considered antifungal, antibacterial, antiviral, and immune modulatory (8) and are able to bind to

DNA(9) Imidazolidine is generally a five-ring non-exocyclic ring (10), Imidazolidine, also known as tetra hydrimidazole, is a bioactive nitrogen containing a non-imported cyclic moiety which provides a wide range of important biological effects such as antioxidants ,antibacterial, antiphlastic, oral anti diabetic activities (11-14), Oxazepine is one of the carbon atoms that contains two oxygen atoms at position (1) and a nitrogen atom located at position (3), in addition to five carbon atoms (15), 1,3-Oxazepine pathways were prepared by closing reactions that result in the condensation of the sheaf residues with different anhydrides such as maleic anhydride and phaleic anhydride(16), Oxazepine derivatives have gained great importance and have been used in a number of fields due to their diverse biological activity, including antibacterial (17), anticancer (18), and antifungal (19), so most of their applications have been in the biological and pharmaceutical fields (20,21). Beta-lactams (2-azetidinones) are saturated heterocyclic compounds

with a four-membered ring that includes three carbon atoms, a nitrogen atom, and a carbonyl group. The name " β -lactam" is given to the cyclic amides because the nitrogen atom is connected to the β carbon atom relative to the carbonyl group (22). β -Lactams, as a structural unit found in the most commonly used antibiotics, have held a fundamental position in medicinal chemistry for about a century. With bacterial resistance to conventional antibiotics through β -lactamases, the need for modern antibiotics is emerging, making the synthesis of newer β -lactams more necessary. In addition to their use as antibiotics, β -lactams are increasingly used as basic compounds for other biologically important molecules (23-25). It has been found that beta-lactams act as inhibitors of acylcholesterol transferase, thrombin inhibitors, inhibitors of human cytomegalovirus protease, matrix metalloprotease inhibitors, cysteine protease inhibitors, and apoptosis inducers. Biological activity is usually linked to the nature of the groups connected to N-1, C-3, and C-4 of the beta-lactam molecules (26). 2-azetidinone derivatives that contain a beta-lactam core have a wide range of pharmacological activity and have become an essential part of the chemical therapeutic arsenal available to medical practitioners today (27).

EXPERIMENTAL

Experimental Apparatus All the chemicals used were of the highest grade and were obtained from Fluka and Merck. For melting point measurements, an Electro-Thermal 9300 apparatus from Melting Point Engineering Ltd, UK, was used. Thin-layer chromatography (TLC) was performed on silica gel, and spots were visualized using iodine vapor. FTIR spectra were obtained using a Shimadzu 8400 Fourier transform infrared spectrophotometer with potassium bromide (KBr) discs, and measurements are reported in cm^{-1} . ^1H NMR and ^{13}C NMR spectra, expressed in ppm, were obtained in DMSO- d_6 as a solvent using an Agilent Varian 300 MHz and 75 MHz spectrometer at the University of Tehran, Iran.

Microwave irradiation was used at 120 W and 280 W.

Combination of Schiff base derivative of compound (S)²⁸

The 6-bromo-2-hydroxy-1-naphthaldehyde substance (0.344 g, 0.002 mole) was crushed well, then layered in a 25 ml ceramic jar, then absolute ethanol was added to it as a solvent, then with the amine colors 2-amino-5-bromo pyrimidine (0.188 g, 0.002 mole), after it was crushed well and mixed. Sufficiently suitable mixture on a homogeneous paste, then irradiate it by placing the ceramic lid in the microwave oven and applying it (1 hr.) at (120 w). Following (TLC) at a ratio of (2:4) and then cooling. The resulting mixture was recrystallized and dried using absolute ethanol

Combination of oxazepine derivatives (S₁, S₂)²⁸.

The compound was prepared by linking the previously prepared Schiff base preparation (0.0008 mol.) with phthalic and maleic anhydride (0.0008 mole and crushing the material well with a mortar to the end of the layer in a ceramic shell, where it was exposed to microwave radiation at 120 watts, with intermittent periods. For a period of 15 minutes, and after the end of the ramification, the laboratory temperature was cooled. The material was consumed with gasoline, and the reaction process was followed using the TLC technique, using the mobile phase (ethanol: dry gasoline) with a ratio of (2: 4), and recrystallization was done using absolute ethanol

Combination of the oxazepane derivative (S₃).

The compound (S₃) was prepared by linking the previously prepared Schiff base preparation (0.0008 mol.), succinic anhydride (0.0008 mole), and crushing the material well with a mortar to the end of the layer in a ceramic shell, where it was exposed to microwave radiation at 120 watts. with intermittent periods. For a period of 15 minutes and after the end of the ramification, the laboratory temperature was cooled. The material was consumed with gasoline, and the reaction process was followed using the TLC technique, using the mobile phase

(ethanol: benzene) with a ratio of (2; 4), and recrystallization was done using absolute ethanol.

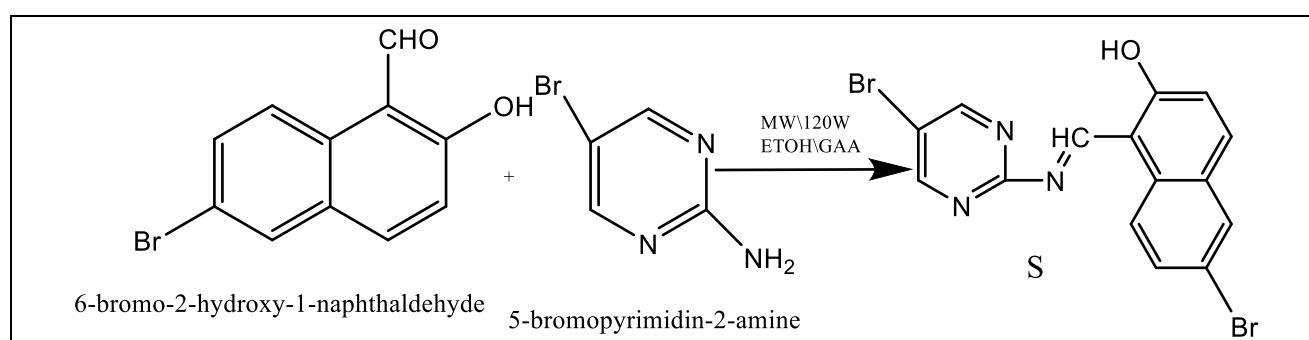
Combination of the Imidazolidine derivatives (S₃).

This compound (0.02 mol, 0.014 g) was mixed with glycine and tyrosine (0.02 mol) separately in tetrahydrofuran as a solvent, and the mixture was heated under a cap in a microwave oven and using it (1 hour) at 120 watts. This was followed by (TLC) at a ratio of (2:4), then cooling. The resulting mixture

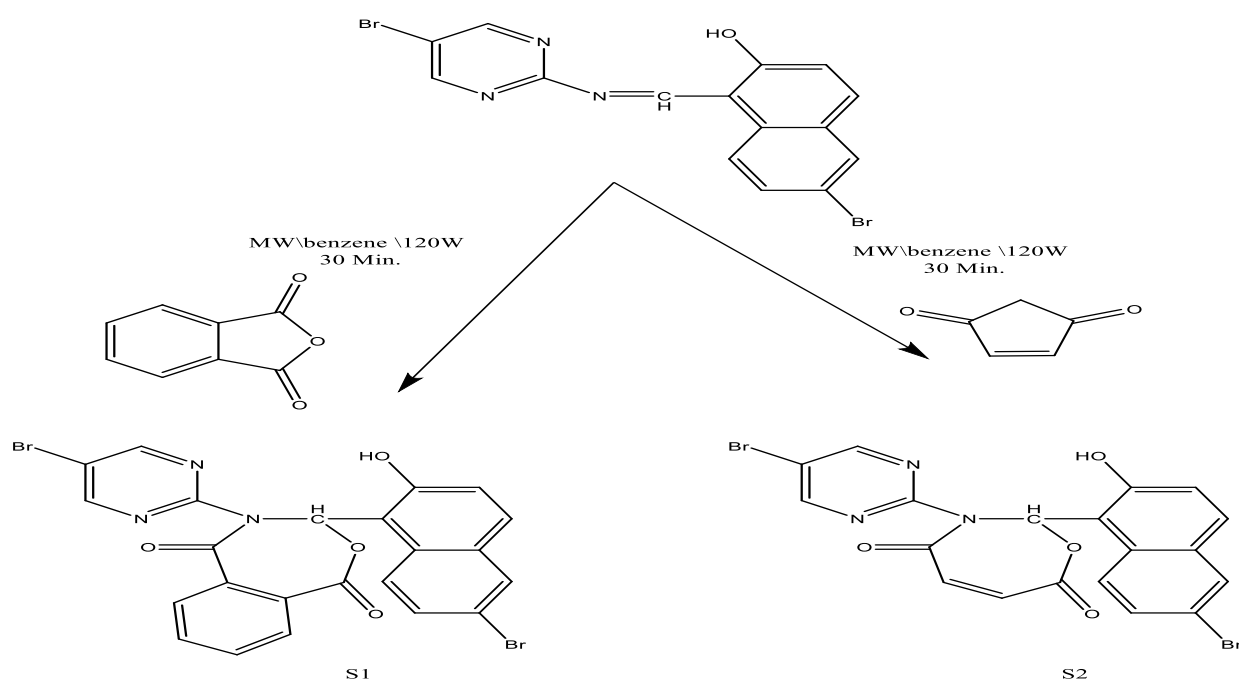
was then dried and crystallized using absolute ethanol.

Combination of β -lactam derivative (S₆)

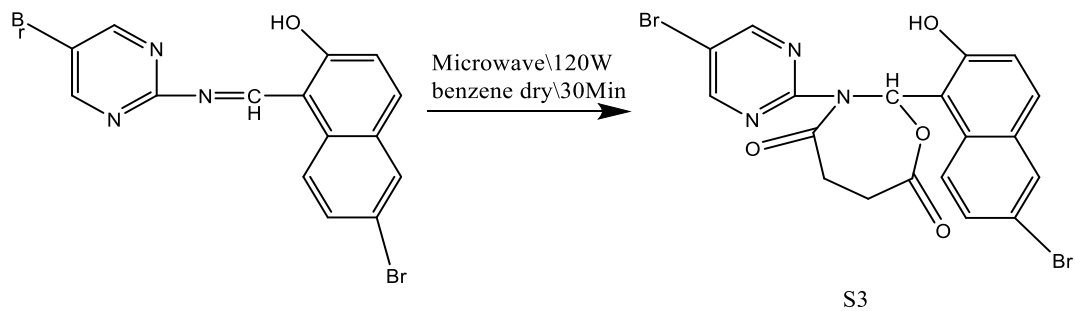
Beta-lactam derivatives (S7) were prepared by reacting compound (S) (0.02 mol) with triethylamine (0.02 mol) and acetyl chloride (0.02 mol) in 1,4-dioxane. The mixture was stirred at 1(15-20) °C for 17 hours to create chemical R6. The products were crystallized from absolute ethanol



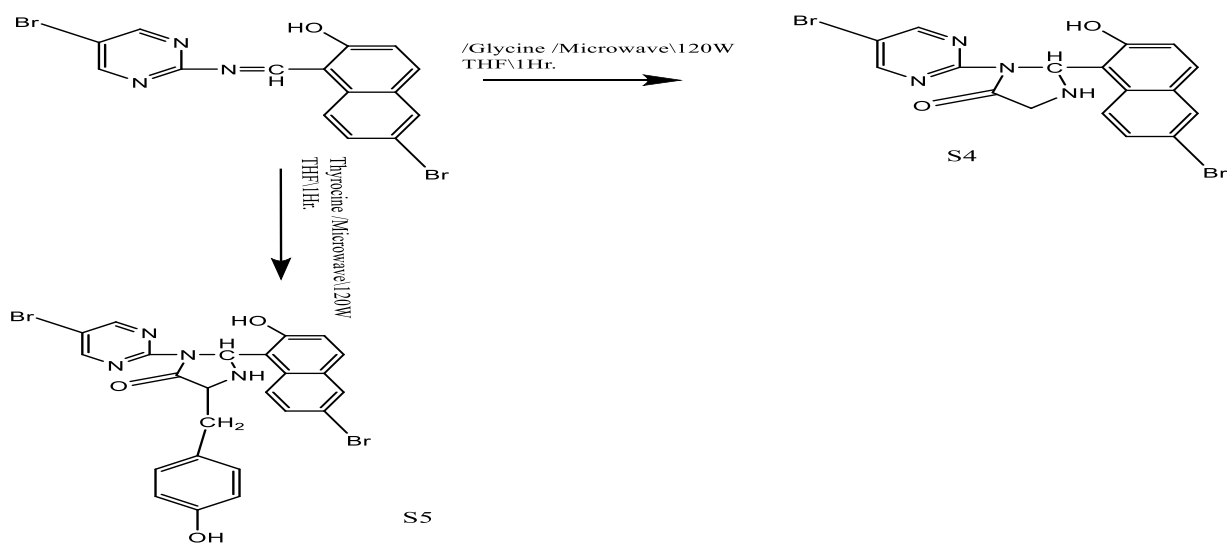
Scheme (1): Combination of compounds (S)



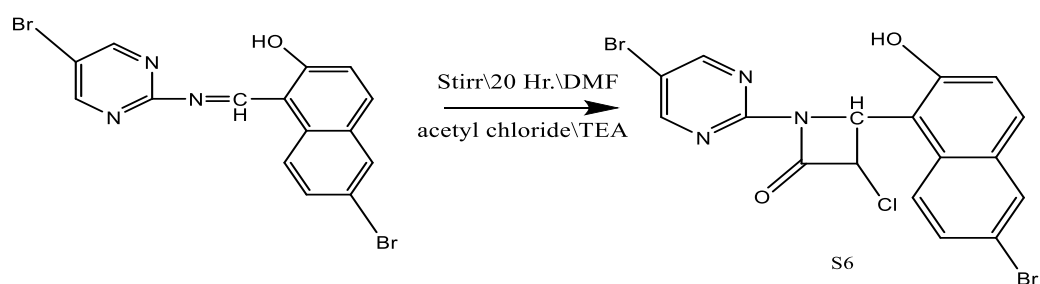
Scheme (2): Combination of compounds (S1, S2)



Scheme (3): Combination of compounds (S3)



Scheme (4): Combination of compounds (S4, S5)



Scheme (5): Combination of compounds (S6)

Study of the biological activity of the compound by paper technique disks (19)

The antibacterial activity on four types of bacteria (*Staphylococcus aureus* and *Escherichia coli*, *Enterococcus faecalis*, *Pseudomonas aeruginosa*) was measured using a filter paper type (Whiteman NO. 1) to prepare 200 tablets after purification. Subsequently, the tablets were placed in test tubes with an average of (5) tablets per tube, and (1 ml) of the synthetic compound solution was added. Weights of (5 mg, 10 mg, 20 mg) of the synthetic compounds were used. The standard used was ciprofloxacin 10. Forceps were sterilized using a flame.

Results and Discussion:

S: (Z)-1-(5-bromo pyrimidin-2-ylimino)methylnaphthalen-2-ol, I.R : , ν 3421(OH), ν 3082($\text{CH}_{\text{aromatic}}$), ν 1657($\text{C}=\text{N}$), ν 1570($\text{C}=\text{C}$)_{aromatic}, (DMSO), δ 11.5 (s, hydroxyl,1H)_{hydroxyl group}, δ 8.3(s, $\text{C}=\text{N}$)_{Imine group}, δ 7.0-7.3(m, Benzene group group, (DMSO), δ 160.486 (C of $\text{C}=\text{N}$)_{pyrimidine ring}, δ 158 (C of $\text{C}=\text{N}$)_{imine group}, δ 156.486 (C of $\text{C}=\text{N}$)_{pyrimidine ring}, δ 153.486 (C of C-OH)_{hydroxyl group} 105-130(C, Benzene group).

S1 : 3-(6-bromo2-hydroxynaphthalen-1-yl)-4-(5-bromo pyrimidin-2-yl)-3,4-dihydrobenzo[e][1,3]oxazepine-1,5-dione, FT-IR(KBr) cm^{-1} , ν 3421(OH), ν 3021($\text{CH}_{\text{aromatic}}$), , ν 1722 (O-C=O)_{lacton}, ν 1687(N-C=O)_{lactam}, ν 1627 ($\text{C}=\text{C}$)_{aromatic}, $^1\text{H-NMR}$ (DMSO), δ 11.7 (s, OH,1H)_{hydroxyl group group}, δ 9.3(s, CH-N)_{Oxazipne ring}, δ 7.0-7.4(m, Benzene group, $^{13}\text{C NMR}$ (DMSO): δ 176.3(C, C=O)_{Lactone ring}, δ 172.1(C, C=O)_{Lactam group}, δ 162 (C of CH-N)_{Oxazepine ring}, δ 160 (C of $\text{C}=\text{N}$)_{imine group}, δ 155 (C of $\text{C}=\text{N}$)_{pyrimidine ring}, δ 153.486 (C of C-OH)_{hydroxyl group} 106-111(C, Benzene group).

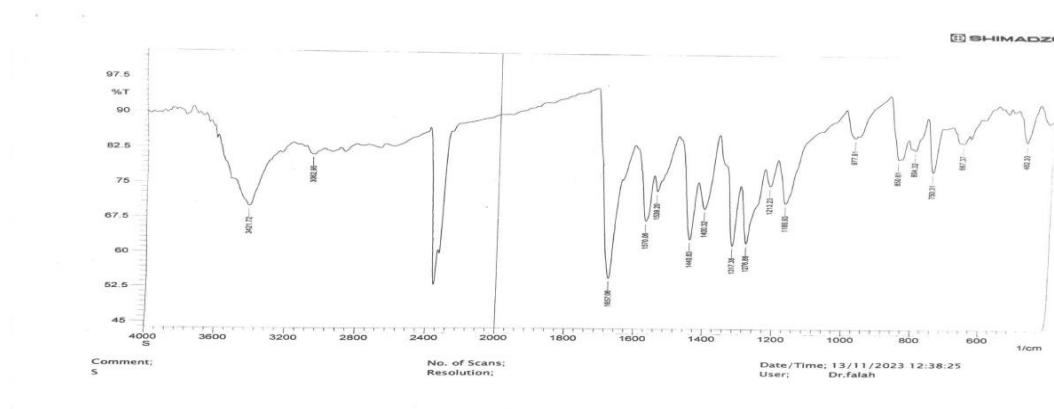
S2: 2-(6-bromo 2-hydroxynaphthalen-1-yl)-3-(5-bromo pyrimidin-2-yl)-2,3-dihydro-1,3-oxazepine-4,7-dione: , I.R (KBr) cm^{-1} , ν 3489(OH), ν 3064($\text{CH}_{\text{aromatic}}$), , ν 1714 (O-C=O)_{lacton}, ν 1680(N-C=O)_{lactam}, , ν 1635 ($\text{C}=\text{C}$)_{oline} ν 1579 ($\text{C}=\text{C}$)_{aromatic} (DMSO), δ 11.2 (s, OH,1H)_{hydroxyl group group}, δ 9.2(s, CH-N)_{Oxazipne ring}, δ 7.0-7.4(m, Benzene group, δ 6.76-6.78 (d,2H, CH_2)_{Methene group}, (DMSO): δ 177(C, C=O)_{Lactone ring}, δ 172.1(C, C=O)_{Lactam group}, δ 162 (C of CH=CH)_{Methene group} δ 161 (C of CH-N)_{Oxazepine ring}, δ 160 (C of $\text{C}=\text{N}$)_{pyrimidine ring}, δ 157 (C of $\text{C}=\text{N}$)_{pyrimidine ring}, δ 155.486 (C of C-OH)_{hydroxyl group} 112-130(C, Benzene group).

S3: 2-(6-bromo2-hydroxynaphthalen-1-yl)-3-(5-bromo pyrimidin-2-yl)-1,3-oxazepane-4,7-dione, I.R (KBr) cm^{-1} , ν 3408(OH), ν 3015($\text{CH}_{\text{aromatic}}$), ν 2927-2864 ($\text{CH}_{\text{aliphatic}}$), ν 1720 (O-C=O)_{lacton}, ν 1685(N-C=O)_{lactam}, ν 1527 ($\text{C}=\text{C}$)_{aromatic}, (DMSO), δ 11.3 (s, OH,1H)_{hydroxyl group group}, δ 9.2(s, CH-N)_{Oxazipne ring}, δ 7.0-7.5(m, Benzene group), δ 1.9 (s, CH_2COO)_{Lactone group}, δ 1.3 (s, CH_2CO)_{Lactam group}, (DMSO): δ 175(C, C=O)_{Lactone ring}, δ 173.1(C, C=O)_{Lactam group}, 161 (C of CH-N)_{Oxazepine ring}, δ 161 (C of $\text{C}=\text{N}$)_{pyrimidine ring}, δ 156 (C of $\text{C}=\text{N}$)_{pyrimidine ring}, δ 152. (C of C-OH) _{hydroxyl group} 121-131(C, Benzene group).

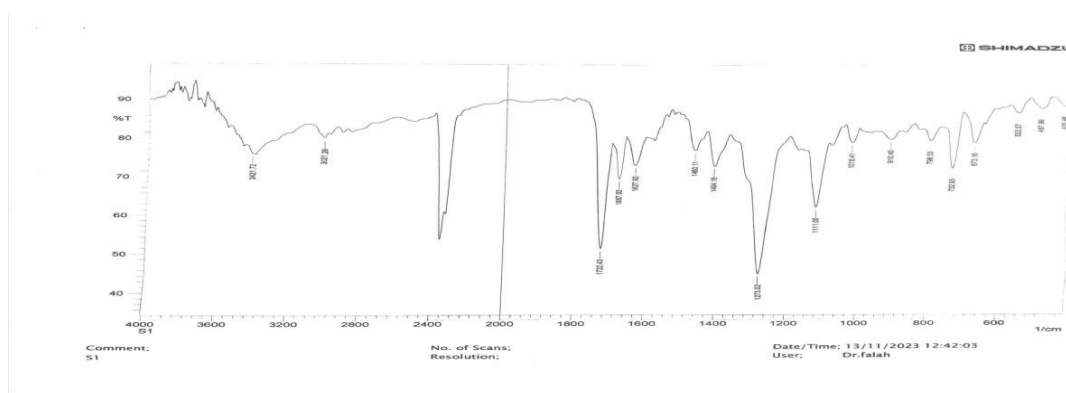
S4: 2-(6-bromo2-hydroxynaphthalen-1-yl)-3-(5-bromopyrimidin-2-yl)imidazolidin-4-one FT-IR(KBr) cm^{-1} , ν 3475(OH) ν 3288($\text{NH}_{\text{imidazolidine ring}}$), ν 3015($\text{CH}_{\text{aromatic}}$), ν 1716($\text{C}=\text{O}$)_{imidazolidine ring}, ν 1624-1579 ($\text{C}=\text{C}$)_{aromatic}, (DMSO), δ 11.4 (s, OH,1H)_{hydroxyl group group}, δ 9.6(s, NH)_{imidazolidine ring}, δ 9.3(s, CH-N)_{imidazolidine ring}, , δ 7.2-7.7(m, Benzene group), δ 4.7 (s, CH_2)_{imidazolidine ring}, (DMSO): δ 175(C, C=O)_{Imidazolidine ring}, δ , 161 (C of CH-N)_{Imidazolidine ring}, δ 160 (C of $\text{C}=\text{N}$)_{pyrimidine ring}, δ 157 (C of $\text{C}=\text{N}$)_{pyrimidine ring}, δ 153. (C of C-OH)_{hydroxyl group} 105-130(C, Benzene group), δ , 63 (C of CH_2)_{Imidazolidine ring}.

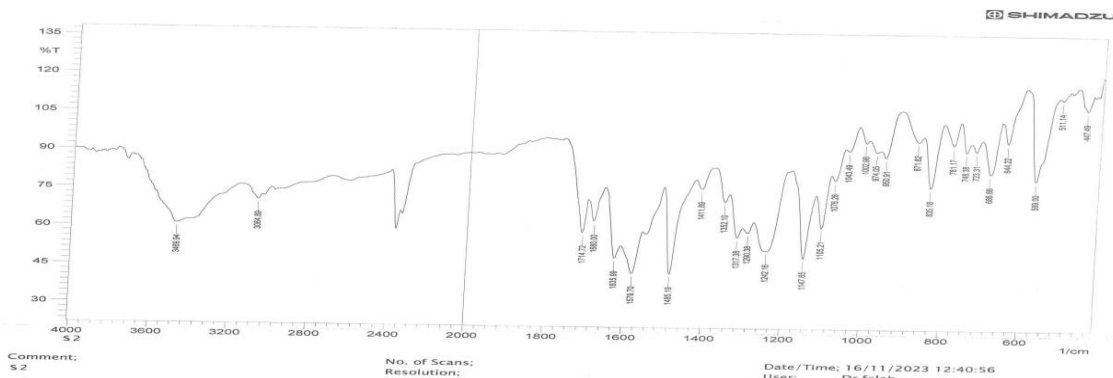
S5: 5-(6-bromo4-hydroxybenzyl)-2-(2-hydroxynaphthalen-1-yl)-3-(5-bromo pyrimidin-2-yl)imidazolidin-4-one FT-IR(KBr) cm^{-1} , ν 3475(OH) ν 3288($\text{NH}_{\text{imidazolidine ring}}$), ν 3015($\text{CH}_{\text{aromatic}}$), ν 1716($\text{C=O}_{\text{imidazolidine ring}}$), ν 1624-1579 ($\text{C=C}_{\text{aromatic}}$), $^1\text{H-NMR}(\text{DMSO})$, δ 11.5 (s, OH, 1H)_{hydroxyl group}, δ 9.8(s, NH)_{imidazolidine ring}, δ 9.3(s, CH-N)_{imidazolidine ring}, δ 7.2-7.7(m, Benzene group), δ 4.32-4.30(trp, 1H)_{imidazolidine ring}, δ 3.3-3.3(d, 2H)_{imidazolidine ring}, $^{13}\text{C NMR}(\text{DMSO})$: δ 172(C, C=O)_{imidazolidine ring}, δ , 162.145 (C of CH-N)_{imidazolidine ring}, δ 159 (C of C=N)_{pyrimidine ring}, δ 155 (C of C-OH)_{hydroxyl group}, δ 151. (C of C-OH)_{hydroxyl group} 105-130(C, Benzene group), δ 62 (C of CH)_{imidazolidine ring} δ 58 (C of CH₂)_{thyrocine ring},

S6: 3-chloro-4-(6-bromo 2-hydroxynaphthalen-1-yl)-1-(5-bromo pyrimidin-2-yl)azetidin-2-one, FT-IR(KBr) cm^{-1} , ν 3408(OH), ν 3064($\text{CH}_{\text{aromatic}}$) ν 1672($\text{C=O}_{\text{lactam}}$), 1597-1548 ($\text{C=C}_{\text{aromatic}}$), ν 800-600(C-Cl), $^1\text{H-NMR}(\text{DMSO})$, δ 11.8 (s, OH, 1H)_{hydroxyl group}, δ 9.4-9.4(d, CH-N)_{lactam ring}, δ 7.0-7.9(m, Benzene group), δ 4.6-4.6(d, 1H)_{lactam ring}, $^{13}\text{C NMR}(\text{DMSO})$: δ 174(C, C=O)_{lactam ring}, δ , 163 (C of CH-N)_{lactam ring}, δ 159 (C of C=N)_{pyrimidine ring}, δ 157 (C of C=N)_{pyrimidine ring}, δ 153 (C of C-OH)_{hydroxyl group} 105-130(C, Benzene group), δ 47 (C of C-Cl)_{lactam ring}

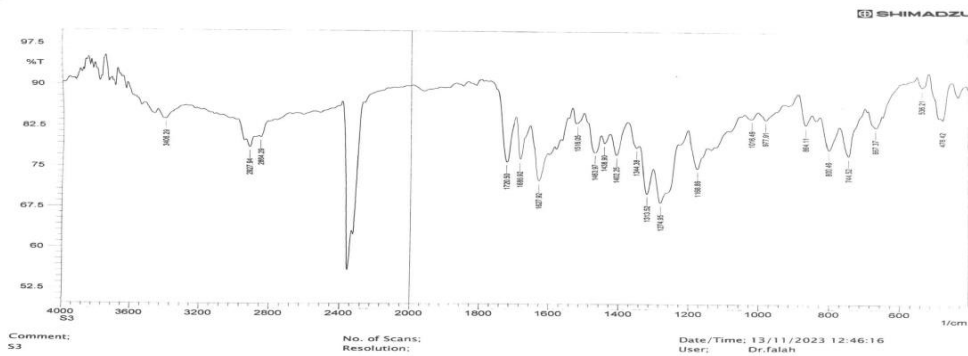


Sym. (1): IR - (S)

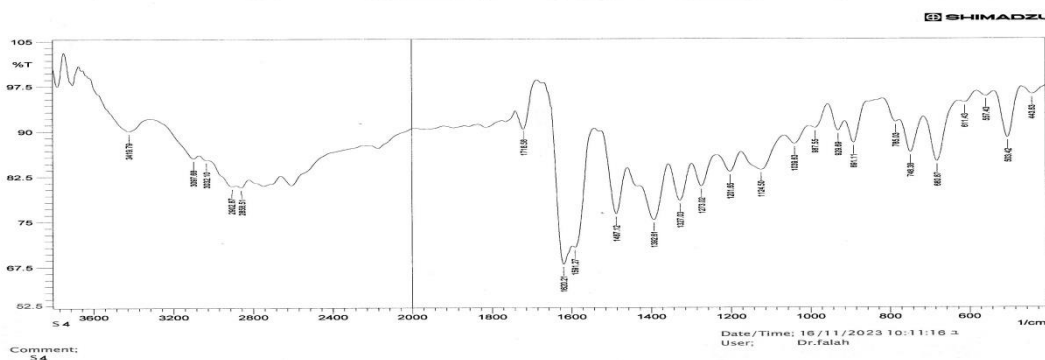
Sym. (2): IR - (S₁)



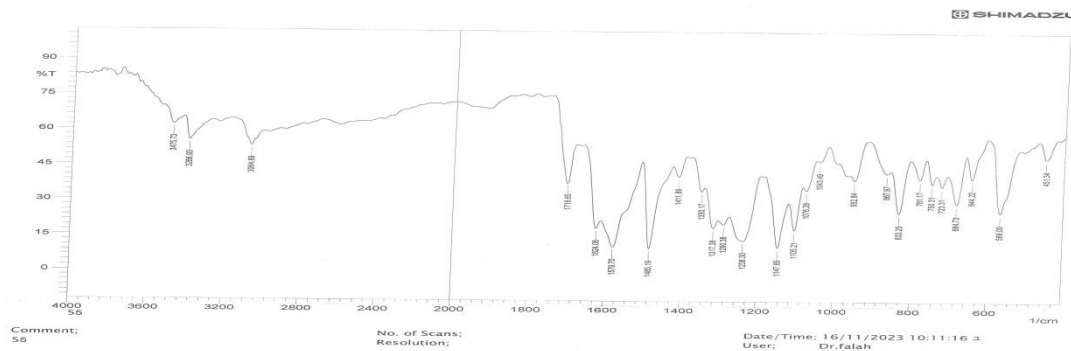
Sym. (3): IR -(S₂)



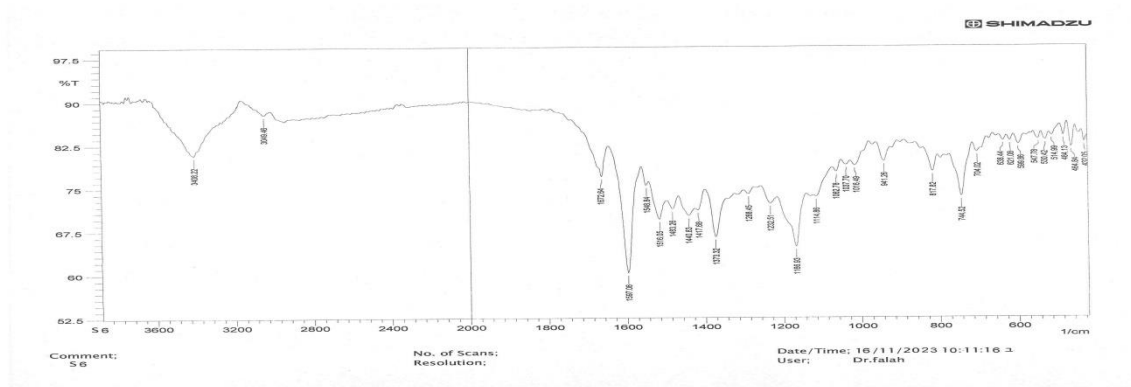
Sym. (4): IR - (S_3)



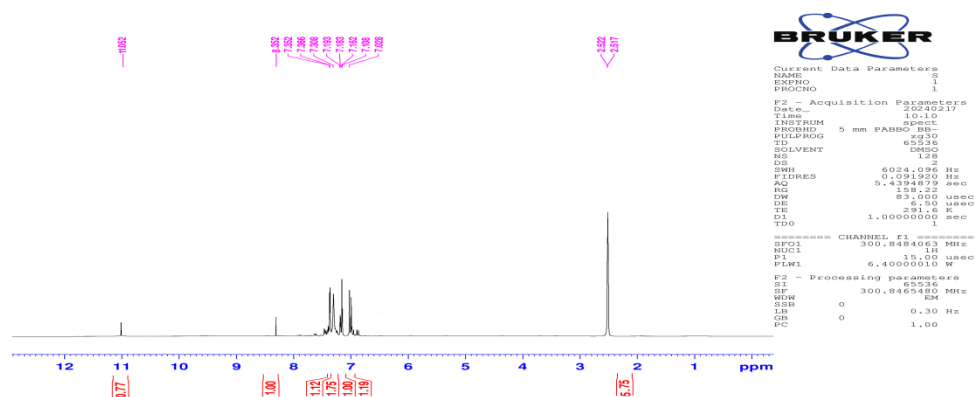
Sym. (5): IR -(S₄)



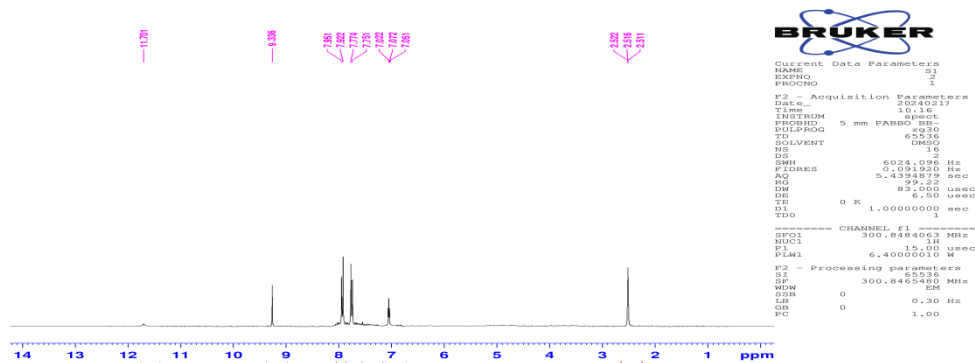
Sym. (6): IR -(S₅)



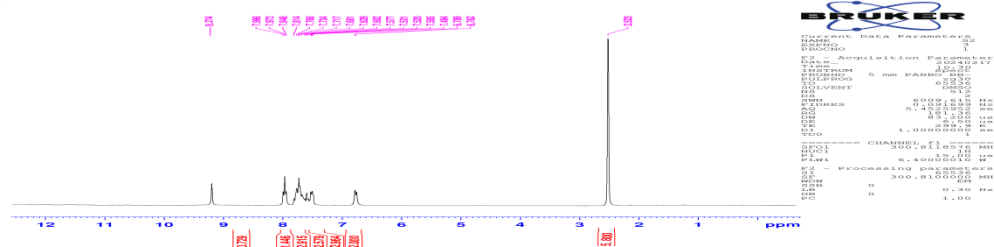
Sym. (7): IR -(S₆)



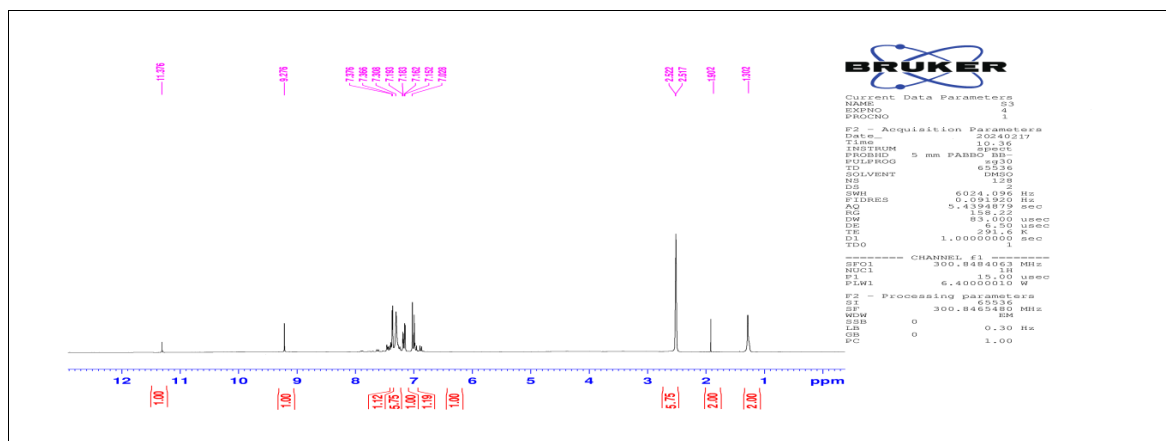
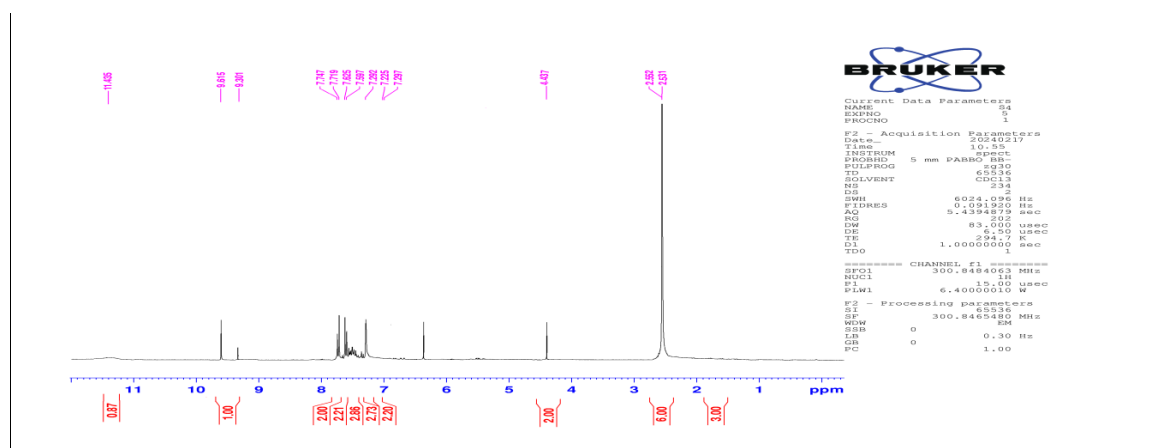
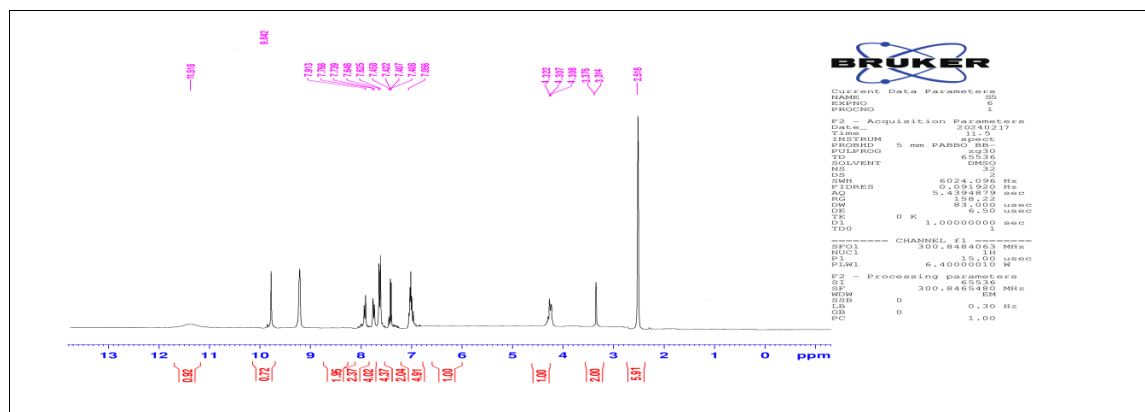
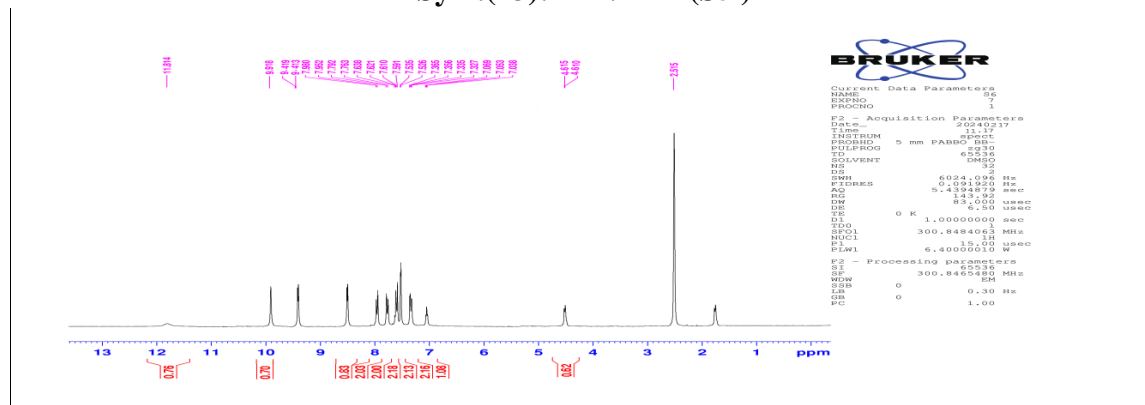
Sym. (8):¹H NMR -(S)

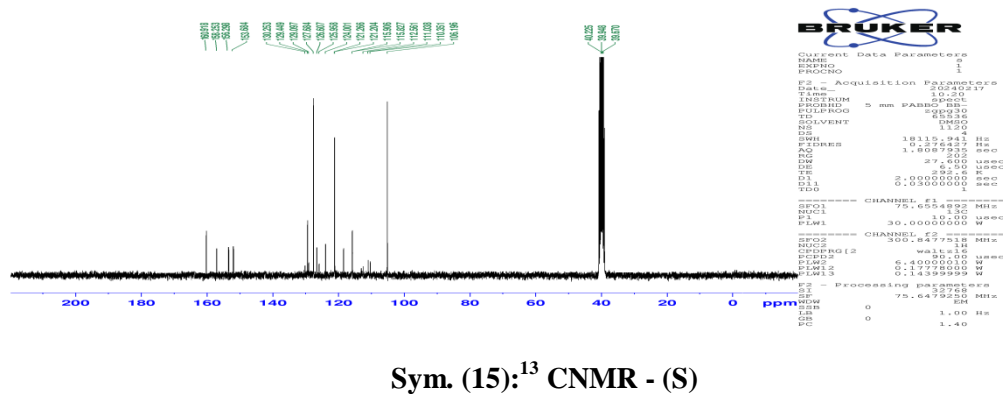


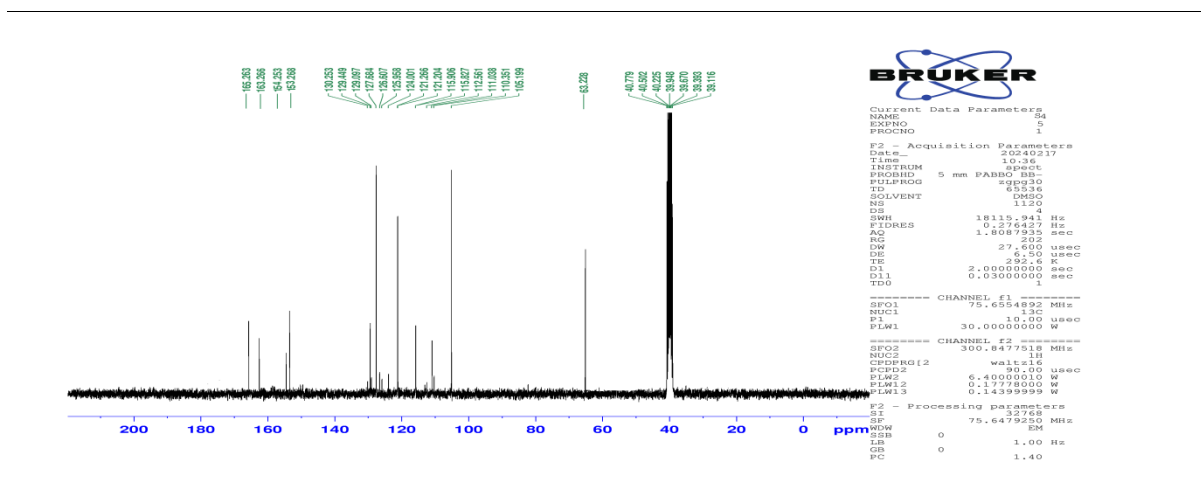
Sym. (9):¹H NMR -(S1)



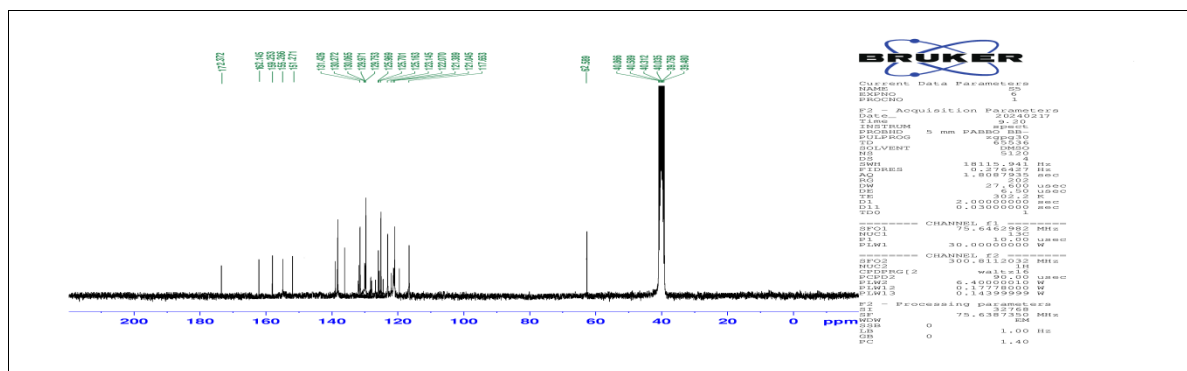
Sym. (10):¹H NMR -(S2)

Sym. (11):¹H NMR - (S3)Sym. (12):¹H NMR - (S4)Sym. (13):¹H NMR - (S5)Sym. (14):¹H NMR - (S6)

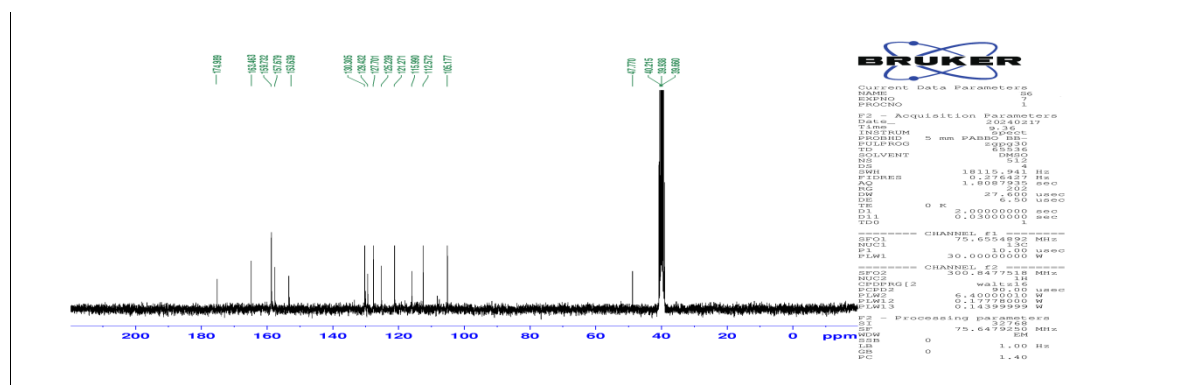




Sym. (19):¹³ CNMR -(S₄)



Sym. (20):¹³ CNMR - (S₅)



Sym. (21):¹³ CNMR -(S₆)

Stall. 1: Physical properties of Combination compounds

product	Struc.	MW	R _F	color	crop%	flush
S	C ₁₅ H ₁₁ Br ₂ N ₃ O	404.9	0.93	yellow	76	ethanol
S ₁	C ₂₃ H ₁₅ Br ₂ N ₃ O ₄	584.9	0.91	orange	76	1,4-dioxane
S ₂	C ₁₉ H ₁₃ Br ₂ N ₃ O ₄	502.9	0.82	Yellow	81	Dry benzene
S ₃	C ₁₉ H ₁₅ N ₃ O ₄	504.9	0.75	yellow	83	Dry benzene
S ₄	C ₁₇ H ₁₄ N ₄ O ₂	589.4	0.79	yellow	83	Dry benzene
S ₅	C ₂₄ H ₂₀ N ₄ O ₃	570.2	0.81	Yellow	88	Tetrahydrofuran
S ₆	C ₁₇ H ₁₂ Br ₂ CIN ₃ O ₂	480.9	0.73	yellow	83	=

Stall 2: Results of Biological Commotion from Types of Bacteria

Sort of bacteria Comp.NO	reserve section (mm) 5mg 10mg 20mg (mg\mol)			
	<i>klebsiella pneumonia</i>	<i>Staphylococcus</i>	<i>Enterococcus faecalis</i>	<i>pseudomonas aeruginosa</i>
S	-, -, 5	-, -, 5	-, -, 5	-, -, 8
S ₁	-	-, -, 4	-, -, 8	-, -, 8
S ₂	-, -, 8	-, 5, 8	-, -, -	-, -, 6
S ₃	-, -, 6	-, -, 10	8, 10, 20	-, -, 10
S ₄	-	-, -, 10	8, 10, -	-, -, 7
S ₅	-	-	8, 10, -	12, 18, 20
S ₆	-	-, 5, 10	-	-



Fig. 1: The antimicrobial activity of compounds (S1-S6)

Conflict of interest: NIL

Funding: NIL

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